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N-(6-Bromomethyl-2-pyridyl)acetamide

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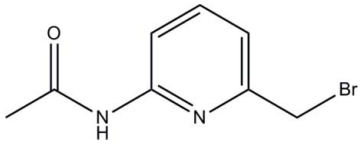
Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.003 Å; R factor = 0.033; wR factor = 0.100; data-to-parameter ratio = 30.1.

The title acetamide compound, $C_8H_9BrN_2O$, crystallizes with three crystallographically independent molecules (*A*, *B* and *C*) in the asymmetric unit. In molecule *A*, the mean plane through the acetamide unit is inclined at a dihedral angle of 4.40 (11)° with respect to the pyridine ring [10.31 (12) and 2.27 (11)°, respectively, for molecules *B* and *C*]. In the crystal structure, molecules are interconnected into sheets parallel to the *ac* plane by N—H···O, C—H···Br, C—H···O and C—H···N hydrogen bonds. The structure is further stabilized by weak intermolecular C—H···π interactions.

Related literature

For general background and applications of acetamide compounds, see: Goswami *et al.* (2000, 2005); Ghosh & Masanta (2006). For the preparation, see: Goswami *et al.* (2001, 2004). For the stability of the temperature controller used in the data collection, see: Cosier & Glazer (1986).



Experimental

Crystal data

$C_8H_9BrN_2O$
M_r = 229.08
Monoclinic, $P2_1/c$
a = 4.1894 (8) Å
b = 26.219 (5) Å
c = 23.817 (4) Å
β = 94.148 (4)°
V = 2609.2 (8) Å^3
Z = 12
Mo $K\alpha$ radiation
μ = 4.68 mm^{-1}
T = 100 K
0.31 × 0.14 × 0.09 mm

Data collection

Bruker APEXII DUO CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)
T_{min} = 0.323, T_{max} = 0.668
72227 measured reflections
10228 independent reflections
8239 reflections with $I > 2\sigma(I)$
R_{int} = 0.058

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.033
$wR(F^2)$ = 0.100
S = 1.06
10228 reflections
340 parameters
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 1.37 e Å^{-3}
$\Delta\rho_{min}$ = −0.74 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*1 and *Cg*2 are the centroids of the C2*A*–C6*A*/N1*A* and C2*C*–C6*C*/N1*C* pyridine rings, respectively.

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N2*A*—H2*NA*···O1*C*^{i}	0.74 (3)	2.29 (3)	3.022 (2)	172 (4)
N2*B*—H2*NB*···O1*A*	0.93 (3)	1.97 (3)	2.885 (2)	166 (3)
N2*C*—H2*NC*···O1*B*^{ii}	0.73 (3)	2.18 (3)	2.900 (2)	169 (3)
C1*B*—H1*BA*···Br1*B*^{iii}	0.97	2.85	3.716 (2)	149
C8*B*—H8*BB*···O1*A*	0.96	2.50	3.159 (3)	125
C8*C*—H8*CA*···N1*A*^{iv}	0.96	2.50	3.427 (3)	162
C1*A*—H1*AB*···*Cg*1^{iii}	0.97	2.88	3.612 (2)	133
C1*C*—H1*CB*···*Cg*2^{iii}	0.97	2.81	3.447 (2)	124

Symmetry codes: (i) $x+1, -y+\frac{1}{2}, z-\frac{1}{2}$; (ii) $x+1, y, z+1$; (iii) $x+1, y, z$; (iv) $x-1, -y+\frac{1}{2}, z+\frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL* software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI5177).

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supplementary materials

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N-(6-Bromomethyl-2-pyridyl)acetamide

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Comment

Pyridine amides having bromine in side chains are enormously useful as they are suitable intermediates for the synthesis of flexible receptors for various biologically important substrates. In addition, they can easily be coupled with alcohol by Williamson reaction and to the amine by a simple reaction with a base. These types of compounds are therefore attracting the attention of molecular recognition chemist (Goswami *et al.*, 2000, 2005; Ghosh & Masanta, 2006).

The title acetamide compound crystallizes in space group $P2_1/c$ with three crystallographically independent molecules in the asymmetric unit, designated *A*, *B* and *C* (Fig. 1). The molecular geometries of all molecules are essentially similar, as indicated by the r.m.s. deviations for the superposition of the non-H atoms of any pair of molecules using *XP* in *SHELXTL* (Sheldrick, 2008) being 0.137 (*A*/*B* pair), 0.026 (*A*/*C* pair) and 0.130 Å (*B*/*C* pair). The superposition of molecular pairs are shown in Fig. 2. The corresponding geometric parameters of the three molecules agree well with each other. In molecule *A*, the mean plane formed through the acetamide moiety (N2A/C7A/C8A/O1A) is inclined at an interplanar angle of 4.40 (11)° with the pyridine ring (C2A-C6A/N1A); the respective angles for molecules *B* and *C* are 10.31 (2) and 2.27 (11)°, respectively.

In the crystal structure, intermolecular N2A—H2NA···O1C, N2B—H2NB···O1A, N2C—H2NC···O1B, C1B—H1BA···Br1B, C8B—H8BB···O1A and C8C—H8CA···N1A hydrogen bonds (Table 1) interconnect molecules into two-molecule-wide arrays parallel to *ac* plane (Fig. 3). Further stabilization of the crystal structure is provided by weak intermolecular C1A—H1AB···*Cg*1 and C1C—H1CB···*Cg2* interactions (Table 1) where *Cg*1 and *Cg*2 are the centroids of C2A-C6A/N1A and C2C-C6C/N1C pyridine rings, respectively.

Experimental

The title compound was prepared according to literature procedures (Goswami *et al.*, 2001, 2004) and was recrystallized from a mixture of $CHCl_3$ and CH_3OH (9:1) by slow evaporation method.

Refinement

H atoms bound to N atoms are located in a difference Fourier map and allowed to refine freely [range of N—H = 0.73 (3)–0.93 (3) Å]. The remaining H atoms were placed in their calculated positions, with C—H = 0.93–0.97 Å, and refined using a riding model, with U_{iso} = 1.2 or 1.5 U_{eq}(C). The rotating group model was applied to methyl groups.

Figures

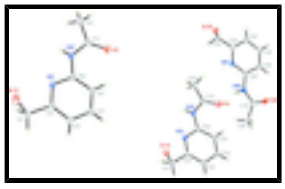


Fig. 1. The molecular structure of the title compound, showing 50% probability displacement ellipsoids for non-H atoms and the atom-numbering scheme.

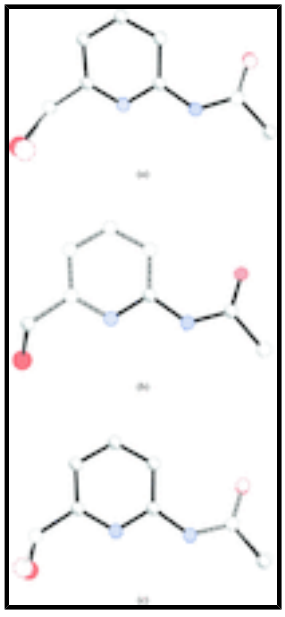


Fig. 2. Fit of (a) molecule *A* (dashed lines) on molecule *B* (solid lines), (b) molecule *C* (dashed lines) on molecule *A* (solid lines), (c) molecule *C* (dashed lines) on molecule *B* (solid lines). H atoms have been omitted for clarity.

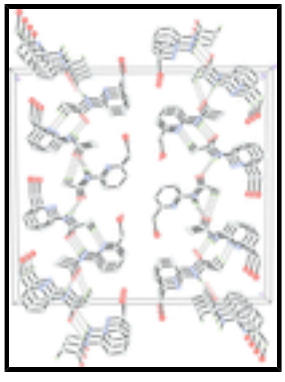


Fig. 3. The crystal structure of the title compound, viewed along the *a* axis. Intermolecular hydrogen bonds are shown as dashed lines.

N-(6-Bromomethyl-2-pyridyl)acetamide

Crystal data

$C_8H_9BrN_2O$	$F(000) = 1368$
$M_r = 229.08$	$D_x = 1.750$ Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 9903 reflections
$a = 4.1894$ (8) Å	$\theta = 3.0$–$33.5°$
$b = 26.219$ (5) Å	$\mu = 4.68$ mm^{-1}

c = 23.817 (4) Å	T = 100 K
β = 94.148 (4)°	Plate, brown
V = 2609.2 (8) Å^3	0.31 × 0.14 × 0.09 mm
Z = 12	

Data collection

Bruker APEXII DUO CCD area-detector diffractometer	10228 independent reflections
Radiation source: fine-focus sealed tube	8239 reflections with $I > 2\sigma(I)$
graphite	R_{int} = 0.058
φ and ω scans	θ_{max} = 33.7°, θ_{min} = 1.8°
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)	h = −6→6
T_{min} = 0.323, T_{max} = 0.668	k = −40→40
72227 measured reflections	l = −36→36

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)]$ = 0.033	Hydrogen site location: inferred from neighbouring sites
$wR(F^2)$ = 0.100	H atoms treated by a mixture of independent and constrained refinement
S = 1.06	$w = 1/[\sigma^2(F_o^2) + (0.052P)^2 + 0.7624P]$ where $P = (F_o^2 + 2F_c^2)/3$
10228 reflections	$(\Delta/\sigma)_{max}$ = 0.005
340 parameters	$\Delta\rho_{max}$ = 1.37 e Å^{-3}
0 restraints	$\Delta\rho_{min}$ = −0.74 e Å^{-3}

Special details

Experimental. The crystal was placed in the cold stream of an Oxford Cryosystems Cobra open-flow nitrogen cryostat (Cosier & Glazer, 1986) operating at 100.0 (1)K.

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > 2\text{sigma}(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Br1A	1.14992 (5)	0.418118 (8)	0.367324 (7)	0.02031 (5)
O1A	0.5948 (4)	0.25161 (6)	0.13677 (6)	0.0296 (3)
N1A	1.1052 (4)	0.35205 (6)	0.24649 (6)	0.0167 (3)
N2A	0.9447 (4)	0.27575 (6)	0.20938 (7)	0.0191 (3)
C1A	1.3223 (5)	0.42675 (8)	0.29327 (7)	0.0196 (3)
H1AA	1.3460	0.4628	0.2856	0.024*
H1AB	1.5326	0.4112	0.2939	0.024*
C2A	1.1091 (4)	0.40309 (7)	0.24723 (7)	0.0166 (3)
C3A	0.9351 (5)	0.43276 (7)	0.20772 (7)	0.0186 (3)
H3AA	0.9413	0.4682	0.2095	0.022*
C4A	0.7516 (5)	0.40783 (7)	0.16540 (7)	0.0200 (3)
H4AA	0.6331	0.4267	0.1382	0.024*
C5A	0.7432 (5)	0.35522 (7)	0.16334 (7)	0.0195 (3)
H5AA	0.6212	0.3381	0.1351	0.023*
C6A	0.9266 (4)	0.32861 (7)	0.20556 (7)	0.0171 (3)
C7A	0.7892 (5)	0.24022 (7)	0.17560 (8)	0.0198 (3)
C8A	0.8755 (5)	0.18609 (7)	0.18922 (8)	0.0235 (4)
H8AA	0.7098	0.1639	0.1735	0.035*
H8AB	0.8985	0.1818	0.2293	0.035*
H8AC	1.0738	0.1778	0.1735	0.035*
Br1B	0.14297 (5)	0.058241 (8)	0.203481 (7)	0.02180 (5)
O1B	0.0701 (5)	0.23172 (6)	−0.04493 (7)	0.0359 (4)
N1B	0.2651 (4)	0.12464 (6)	0.08249 (6)	0.0174 (3)
N2B	0.2563 (4)	0.20320 (6)	0.04135 (7)	0.0206 (3)
C1B	0.3307 (5)	0.04503 (8)	0.13130 (8)	0.0209 (3)
H1BA	0.5573	0.0530	0.1351	0.025*
H1BB	0.3086	0.0091	0.1221	0.025*
C2B	0.1736 (4)	0.07590 (7)	0.08467 (7)	0.0170 (3)
C3B	−0.0419 (5)	0.05349 (7)	0.04525 (8)	0.0199 (3)
H3BA	−0.1004	0.0194	0.0482	0.024*
C4B	−0.1672 (5)	0.08373 (8)	0.00115 (8)	0.0215 (4)
H4BA	−0.3132	0.0700	−0.0260	0.026*
C5B	−0.0763 (5)	0.13402 (8)	−0.00257 (8)	0.0209 (3)
H5BA	−0.1573	0.1546	−0.0321	0.025*
C6B	0.1427 (5)	0.15320 (7)	0.03964 (7)	0.0174 (3)
C7B	0.2223 (6)	0.23918 (8)	0.00022 (8)	0.0251 (4)
C8B	0.3835 (7)	0.28927 (9)	0.01365 (10)	0.0352 (5)
H8BA	0.4373	0.3054	−0.0206	0.053*
H8BB	0.5750	0.2836	0.0375	0.053*
H8BC	0.2414	0.3109	0.0327	0.053*
Br1C	1.07629 (5)	0.426384 (8)	1.034127 (8)	0.02481 (6)
O1C	0.3915 (4)	0.28054 (6)	0.79326 (6)	0.0243 (3)
N1C	0.9830 (4)	0.36463 (6)	0.90997 (6)	0.0173 (3)
N2C	0.7488 (4)	0.29499 (6)	0.86856 (7)	0.0201 (3)
C1C	1.2666 (5)	0.43125 (8)	0.96116 (8)	0.0218 (3)

H1CA	1.3315	0.4662	0.9550	0.026*
H1CB	1.4563	0.4100	0.9619	0.026*
C2C	1.0370 (5)	0.41486 (7)	0.91382 (7)	0.0174 (3)
C3C	0.8994 (5)	0.44982 (7)	0.87597 (7)	0.0202 (3)
H3CA	0.9409	0.4845	0.8800	0.024*
C4C	0.6977 (5)	0.43147 (8)	0.83191 (7)	0.0209 (3)
H4CA	0.6031	0.4541	0.8056	0.025*
C5C	0.6360 (5)	0.37991 (7)	0.82678 (7)	0.0194 (3)
H5CA	0.5004	0.3671	0.7975	0.023*
C6C	0.7858 (4)	0.34775 (7)	0.86741 (7)	0.0168 (3)
C7C	0.5659 (5)	0.26431 (7)	0.83320 (7)	0.0196 (3)
C8C	0.5926 (6)	0.20884 (8)	0.84701 (9)	0.0271 (4)
H8CA	0.5007	0.1892	0.8159	0.041*
H8CB	0.4802	0.2018	0.8799	0.041*
H8CC	0.8140	0.1999	0.8541	0.041*
H2NA	1.062 (8)	0.2645 (13)	0.2303 (13)	0.045 (9)*
H2NB	0.372 (8)	0.2135 (12)	0.0743 (12)	0.039 (8)*
H2NC	0.850 (8)	0.2790 (12)	0.8879 (12)	0.033 (8)*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Br1A	0.02104 (9)	0.02320 (9)	0.01647 (8)	−0.00081 (6)	−0.00005 (6)	−0.00327 (6)
O1A	0.0389 (9)	0.0208 (7)	0.0268 (7)	−0.0026 (6)	−0.0141 (6)	−0.0012 (6)
N1A	0.0182 (7)	0.0163 (7)	0.0152 (6)	−0.0005 (5)	−0.0012 (5)	−0.0013 (5)
N2A	0.0228 (8)	0.0159 (7)	0.0177 (7)	0.0005 (6)	−0.0055 (6)	−0.0011 (5)
C1A	0.0198 (8)	0.0209 (8)	0.0181 (7)	−0.0055 (6)	0.0007 (6)	−0.0008 (6)
C2A	0.0166 (8)	0.0171 (7)	0.0163 (7)	−0.0012 (6)	0.0014 (6)	−0.0006 (6)
C3A	0.0233 (9)	0.0161 (7)	0.0163 (7)	0.0004 (6)	0.0015 (6)	0.0022 (6)
C4A	0.0235 (9)	0.0201 (8)	0.0163 (7)	0.0014 (7)	−0.0007 (6)	0.0018 (6)
C5A	0.0238 (9)	0.0194 (8)	0.0146 (7)	0.0004 (6)	−0.0032 (6)	−0.0006 (6)
C6A	0.0187 (8)	0.0175 (8)	0.0148 (7)	0.0001 (6)	−0.0005 (6)	−0.0019 (6)
C7A	0.0234 (9)	0.0171 (8)	0.0186 (7)	−0.0017 (6)	−0.0007 (6)	−0.0022 (6)
C8A	0.0287 (10)	0.0177 (8)	0.0236 (8)	−0.0001 (7)	−0.0020 (7)	−0.0032 (7)
Br1B	0.02550 (10)	0.02243 (10)	0.01691 (8)	0.00067 (7)	−0.00224 (6)	0.00250 (6)
O1B	0.0555 (11)	0.0220 (7)	0.0271 (7)	−0.0039 (7)	−0.0182 (7)	0.0056 (6)
N1B	0.0190 (7)	0.0172 (7)	0.0156 (6)	0.0021 (5)	−0.0022 (5)	−0.0002 (5)
N2B	0.0279 (8)	0.0166 (7)	0.0163 (6)	0.0006 (6)	−0.0051 (6)	−0.0002 (5)
C1B	0.0214 (9)	0.0194 (8)	0.0218 (8)	0.0032 (6)	0.0007 (7)	0.0027 (6)
C2B	0.0175 (8)	0.0177 (8)	0.0157 (7)	0.0022 (6)	0.0008 (6)	0.0002 (6)
C3B	0.0219 (9)	0.0194 (8)	0.0186 (7)	−0.0021 (7)	0.0020 (6)	−0.0033 (6)
C4B	0.0225 (9)	0.0252 (9)	0.0165 (7)	−0.0013 (7)	−0.0018 (6)	−0.0035 (6)
C5B	0.0224 (9)	0.0234 (9)	0.0161 (7)	0.0021 (7)	−0.0032 (6)	−0.0008 (6)
C6B	0.0202 (8)	0.0165 (7)	0.0154 (7)	0.0015 (6)	−0.0005 (6)	−0.0010 (6)
C7B	0.0341 (11)	0.0175 (8)	0.0224 (8)	0.0024 (7)	−0.0064 (7)	0.0018 (7)
C8B	0.0512 (15)	0.0194 (9)	0.0323 (11)	−0.0037 (9)	−0.0145 (10)	0.0049 (8)
Br1C	0.02540 (10)	0.03212 (11)	0.01646 (8)	0.00276 (7)	−0.00162 (7)	−0.00555 (7)
O1C	0.0274 (8)	0.0223 (7)	0.0217 (6)	−0.0020 (5)	−0.0090 (5)	0.0002 (5)

N1C	0.0183 (7)	0.0182 (7)	0.0150 (6)	−0.0008 (5)	−0.0012 (5)	−0.0007 (5)
N2C	0.0254 (8)	0.0161 (7)	0.0175 (7)	−0.0022 (6)	−0.0071 (6)	0.0018 (5)
C1C	0.0219 (9)	0.0223 (8)	0.0210 (8)	−0.0039 (7)	−0.0010 (6)	−0.0019 (7)
C2C	0.0189 (8)	0.0189 (8)	0.0145 (7)	−0.0011 (6)	0.0013 (6)	−0.0009 (6)
C3C	0.0260 (9)	0.0181 (8)	0.0162 (7)	−0.0017 (7)	−0.0004 (6)	−0.0011 (6)
C4C	0.0262 (9)	0.0208 (8)	0.0152 (7)	0.0015 (7)	−0.0025 (6)	0.0032 (6)
C5C	0.0221 (9)	0.0203 (8)	0.0151 (7)	0.0008 (6)	−0.0031 (6)	0.0006 (6)
C6C	0.0186 (8)	0.0174 (8)	0.0142 (7)	0.0000 (6)	−0.0009 (6)	−0.0006 (6)
C7C	0.0210 (9)	0.0198 (8)	0.0177 (7)	−0.0025 (6)	−0.0012 (6)	−0.0019 (6)
C8C	0.0355 (12)	0.0187 (9)	0.0254 (9)	−0.0045 (8)	−0.0099 (8)	0.0008 (7)

Geometric parameters (Å, °)

Br1A—C1A	1.9667 (18)	C3B—C4B	1.389 (3)
O1A—C7A	1.224 (2)	C3B—H3BA	0.93
N1A—C6A	1.335 (2)	C4B—C5B	1.377 (3)
N1A—C2A	1.338 (2)	C4B—H4BA	0.93
N2A—C7A	1.365 (2)	C5B—C6B	1.404 (3)
N2A—C6A	1.391 (2)	C5B—H5BA	0.93
N2A—H2NA	0.74 (3)	C7B—C8B	1.501 (3)
C1A—C2A	1.498 (3)	C8B—H8BA	0.96
C1A—H1AA	0.97	C8B—H8BB	0.96
C1A—H1AB	0.97	C8B—H8BC	0.96
C2A—C3A	1.387 (3)	Br1C—C1C	1.968 (2)
C3A—C4A	1.386 (3)	O1C—C7C	1.233 (2)
C3A—H3AA	0.93	N1C—C6C	1.336 (2)
C4A—C5A	1.380 (3)	N1C—C2C	1.338 (2)
C4A—H4AA	0.93	N2C—C7C	1.360 (2)
C5A—C6A	1.406 (2)	N2C—C6C	1.392 (2)
C5A—H5AA	0.93	N2C—H2NC	0.73 (3)
C7A—C8A	1.494 (3)	C1C—C2C	1.491 (3)
C8A—H8AA	0.96	C1C—H1CA	0.97
C8A—H8AB	0.96	C1C—H1CB	0.97
C8A—H8AC	0.96	C2C—C3C	1.382 (3)
Br1B—C1B	1.9722 (19)	C3C—C4C	1.385 (3)
O1B—C7B	1.225 (2)	C3C—H3CA	0.93
N1B—C2B	1.336 (2)	C4C—C5C	1.380 (3)
N1B—C6B	1.338 (2)	C4C—H4CA	0.93
N2B—C7B	1.360 (2)	C5C—C6C	1.398 (2)
N2B—C6B	1.394 (2)	C5C—H5CA	0.93
N2B—H2NB	0.93 (3)	C7C—C8C	1.494 (3)
C1B—C2B	1.489 (3)	C8C—H8CA	0.96
C1B—H1BA	0.97	C8C—H8CB	0.96
C1B—H1BB	0.97	C8C—H8CC	0.96
C2B—C3B	1.385 (3)		
C6A—N1A—C2A	118.40 (15)	C5B—C4B—H4BA	119.9
C7A—N2A—C6A	128.33 (16)	C3B—C4B—H4BA	119.9
C7A—N2A—H2NA	113 (3)	C4B—C5B—C6B	117.78 (17)
C6A—N2A—H2NA	118 (3)	C4B—C5B—H5BA	121.1

C2A—C1A—Br1A	111.74 (13)	C6B—C5B—H5BA	121.1
C2A—C1A—H1AA	109.3	N1B—C6B—N2B	113.14 (16)
Br1A—C1A—H1AA	109.3	N1B—C6B—C5B	122.70 (17)
C2A—C1A—H1AB	109.3	N2B—C6B—C5B	124.16 (16)
Br1A—C1A—H1AB	109.3	O1B—C7B—N2B	122.83 (19)
H1AA—C1A—H1AB	107.9	O1B—C7B—C8B	121.62 (19)
N1A—C2A—C3A	123.14 (17)	N2B—C7B—C8B	115.54 (17)
N1A—C2A—C1A	115.46 (16)	C7B—C8B—H8BA	109.5
C3A—C2A—C1A	121.39 (17)	C7B—C8B—H8BB	109.5
C4A—C3A—C2A	117.74 (17)	H8BA—C8B—H8BB	109.5
C4A—C3A—H3AA	121.1	C7B—C8B—H8BC	109.5
C2A—C3A—H3AA	121.1	H8BA—C8B—H8BC	109.5
C5A—C4A—C3A	120.55 (17)	H8BB—C8B—H8BC	109.5
C5A—C4A—H4AA	119.7	C6C—N1C—C2C	118.06 (16)
C3A—C4A—H4AA	119.7	C7C—N2C—C6C	129.32 (16)
C4A—C5A—C6A	117.35 (17)	C7C—N2C—H2NC	109 (2)
C4A—C5A—H5AA	121.3	C6C—N2C—H2NC	121 (2)
C6A—C5A—H5AA	121.3	C2C—C1C—Br1C	111.62 (13)
N1A—C6A—N2A	112.72 (16)	C2C—C1C—H1CA	109.3
N1A—C6A—C5A	122.82 (17)	Br1C—C1C—H1CA	109.3
N2A—C6A—C5A	124.46 (16)	C2C—C1C—H1CB	109.3
O1A—C7A—N2A	122.80 (18)	Br1C—C1C—H1CB	109.3
O1A—C7A—C8A	122.20 (17)	H1CA—C1C—H1CB	108.0
N2A—C7A—C8A	114.99 (17)	N1C—C2C—C3C	123.20 (17)
C7A—C8A—H8AA	109.5	N1C—C2C—C1C	115.60 (16)
C7A—C8A—H8AB	109.5	C3C—C2C—C1C	121.18 (17)
H8AA—C8A—H8AB	109.5	C2C—C3C—C4C	117.81 (18)
C7A—C8A—H8AC	109.5	C2C—C3C—H3CA	121.1
H8AA—C8A—H8AC	109.5	C4C—C3C—H3CA	121.1
H8AB—C8A—H8AC	109.5	C5C—C4C—C3C	120.48 (17)
C2B—N1B—C6B	118.02 (16)	C5C—C4C—H4CA	119.8
C7B—N2B—C6B	127.83 (16)	C3C—C4C—H4CA	119.8
C7B—N2B—H2NB	115.3 (19)	C4C—C5C—C6C	117.26 (17)
C6B—N2B—H2NB	116.8 (19)	C4C—C5C—H5CA	121.4
C2B—C1B—Br1B	111.85 (13)	C6C—C5C—H5CA	121.4
C2B—C1B—H1BA	109.2	N1C—C6C—N2C	112.20 (16)
Br1B—C1B—H1BA	109.2	N1C—C6C—C5C	123.19 (17)
C2B—C1B—H1BB	109.2	N2C—C6C—C5C	124.62 (16)
Br1B—C1B—H1BB	109.2	O1C—C7C—N2C	123.35 (18)
H1BA—C1B—H1BB	107.9	O1C—C7C—C8C	122.47 (17)
N1B—C2B—C3B	123.66 (17)	N2C—C7C—C8C	114.18 (16)
N1B—C2B—C1B	115.86 (16)	C7C—C8C—H8CA	109.5
C3B—C2B—C1B	120.41 (17)	C7C—C8C—H8CB	109.5
C2B—C3B—C4B	117.55 (18)	H8CA—C8C—H8CB	109.5
C2B—C3B—H3BA	121.2	C7C—C8C—H8CC	109.5
C4B—C3B—H3BA	121.2	H8CA—C8C—H8CC	109.5
C5B—C4B—C3B	120.29 (18)	H8CB—C8C—H8CC	109.5
C6A—N1A—C2A—C3A	0.4 (3)	C2B—N1B—C6B—N2B	−179.55 (16)
C6A—N1A—C2A—C1A	−178.19 (16)	C2B—N1B—C6B—C5B	0.1 (3)

Br1A—C1A—C2A—N1A	−71.41 (19)	C7B—N2B—C6B—N1B	168.3 (2)
Br1A—C1A—C2A—C3A	110.00 (18)	C7B—N2B—C6B—C5B	−11.3 (3)
N1A—C2A—C3A—C4A	−0.5 (3)	C4B—C5B—C6B—N1B	0.3 (3)
C1A—C2A—C3A—C4A	177.93 (17)	C4B—C5B—C6B—N2B	179.93 (19)
C2A—C3A—C4A—C5A	0.3 (3)	C6B—N2B—C7B—O1B	2.6 (4)
C3A—C4A—C5A—C6A	0.1 (3)	C6B—N2B—C7B—C8B	−177.6 (2)
C2A—N1A—C6A—N2A	179.68 (16)	C6C—N1C—C2C—C3C	−0.1 (3)
C2A—N1A—C6A—C5A	0.0 (3)	C6C—N1C—C2C—C1C	−178.23 (17)
C7A—N2A—C6A—N1A	178.45 (19)	Br1C—C1C—C2C—N1C	−73.37 (19)
C7A—N2A—C6A—C5A	−1.9 (3)	Br1C—C1C—C2C—C3C	108.44 (18)
C4A—C5A—C6A—N1A	−0.2 (3)	N1C—C2C—C3C—C4C	−0.4 (3)
C4A—C5A—C6A—N2A	−179.84 (18)	C1C—C2C—C3C—C4C	177.65 (18)
C6A—N2A—C7A—O1A	−2.9 (3)	C2C—C3C—C4C—C5C	0.6 (3)
C6A—N2A—C7A—C8A	176.72 (18)	C3C—C4C—C5C—C6C	−0.3 (3)
C6B—N1B—C2B—C3B	−0.4 (3)	C2C—N1C—C6C—N2C	−179.58 (17)
C6B—N1B—C2B—C1B	176.64 (16)	C2C—N1C—C6C—C5C	0.4 (3)
Br1B—C1B—C2B—N1B	78.77 (19)	C7C—N2C—C6C—N1C	178.90 (19)
Br1B—C1B—C2B—C3B	−104.14 (18)	C7C—N2C—C6C—C5C	−1.1 (3)
N1B—C2B—C3B—C4B	0.2 (3)	C4C—C5C—C6C—N1C	−0.2 (3)
C1B—C2B—C3B—C4B	−176.68 (17)	C4C—C5C—C6C—N2C	179.76 (19)
C2B—C3B—C4B—C5B	0.3 (3)	C6C—N2C—C7C—O1C	−1.6 (3)
C3B—C4B—C5B—C6B	−0.5 (3)	C6C—N2C—C7C—C8C	178.7 (2)

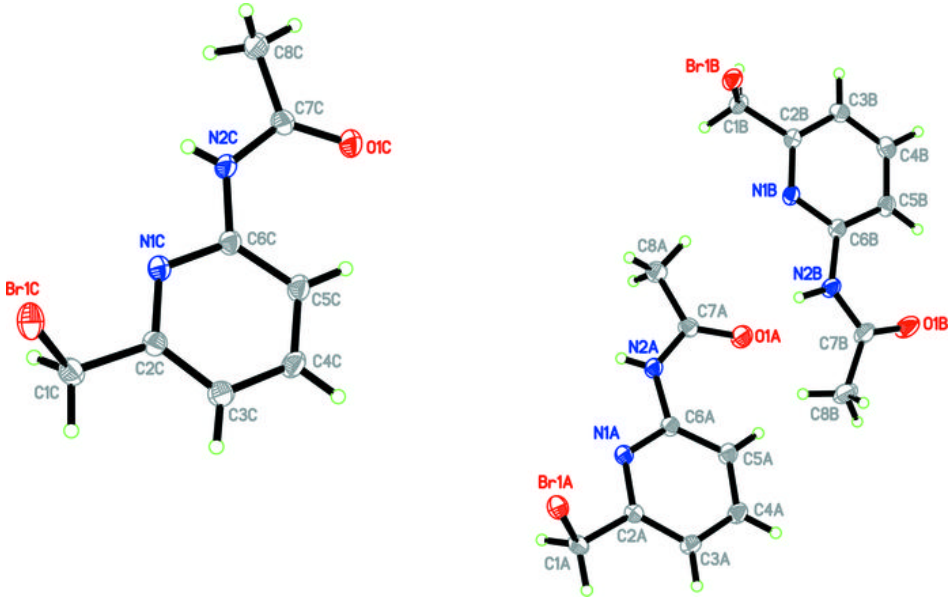
Hydrogen-bond geometry (Å, °)

Cg1 and Cg2 are the centroids of the C2A–C6A/N1A and C2C–C6C/N1C pyridine rings, respectively.

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N2A—H2NA···O1C^{i}	0.74 (3)	2.29 (3)	3.022 (2)	172 (4)
N2B—H2NB···O1A	0.93 (3)	1.97 (3)	2.885 (2)	166 (3)
N2C—H2NC···O1B^{ii}	0.73 (3)	2.18 (3)	2.900 (2)	169 (3)
C1B—H1BA···Br1B^{iii}	0.97	2.85	3.716 (2)	149
C8B—H8BB···O1A	0.96	2.50	3.159 (3)	125
C8C—H8CA···N1A^{iv}	0.96	2.50	3.427 (3)	162
C1A—H1AB···Cg1^{iii}	0.97	2.88	3.612 (2)	133
C1C—H1CB···Cg2^{iii}	0.97	2.81	3.447 (2)	124

Symmetry codes: (i) *x*+1, −*y*+1/2, *z*−1/2; (ii) *x*+1, *y*, *z*+1; (iii) *x*+1, *y*, *z*; (iv) *x*−1, −*y*+1/2, *z*+1/2.

Fig. 1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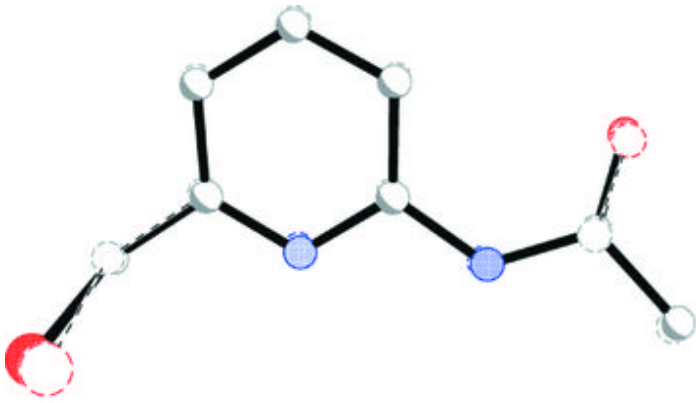
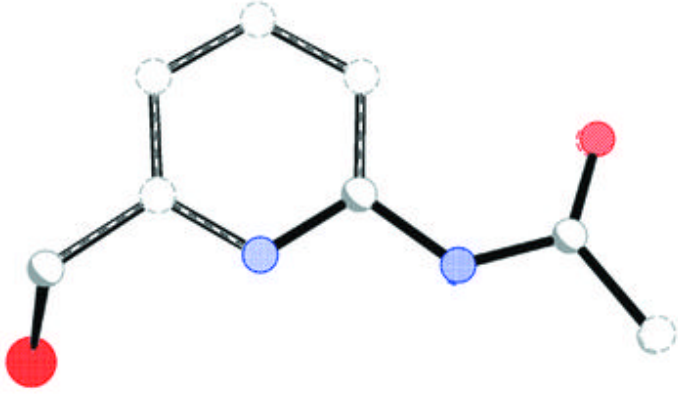
C7C
N2C
O1C
N1C
C6C
Br1C
C5C
C2C
C4C
C1C
C3C
Br1B
C1B
C2B
C3B
C4B
N1B
C5B
C6B
C8A
N2B
C7A
O1A
N2A
C7B
O1B
C8B
C6A
N1A
C5A
Br1A
C4A
C2A
C1A
C3A

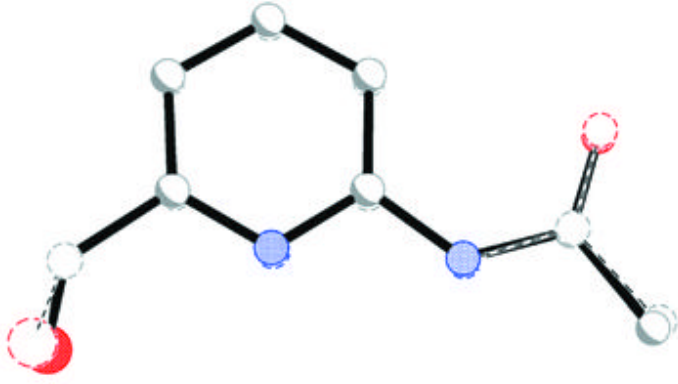
Fig. 2



(a)



(b)



(c)

Fig. 3